


Mid-infrared emission properties of Er³⁺, Dy³⁺-codoped CaF₂ crystals

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ABSTRACT

We present a comprehensive spectroscopic investigation of singly Dy³⁺-doped and Er³⁺, Dy³⁺-co-doped calcium fluoride (CaF₂) crystals for mid-infrared laser applications around 3 μm and 4.4 μm. The *f-f* transition probabilities of Dy³⁺ were evaluated using a modified Judd-Ofelt formalism that accounts for configuration interaction effects. The stimulated-emission cross section (σ_{SE}) reaches $0.25 \times 10^{-20} \text{ cm}^2$ at 2.93 μm for the ⁶H_{13/2} → ⁶H_{15/2} transition, with an emission bandwidth of ~350 nm. This transition exhibits a pronounced phonon sideband extending up to 3.9 μm. The luminescence lifetime of the ⁶H_{13/2} level decreases from 2.04 to 0.52 ms as the Dy³⁺ concentration raises from 0.1 to 0.5 at.%. Energy transfer from Er³⁺ to Dy³⁺ in codoped crystals is quantified, demonstrating an efficient pathway for sensitized pumping. For the first time, 4.4 μm emission from Dy³⁺:CaF₂ is observed at room temperature. The interplay between radiative and non-radiative transition rates is analyzed, highlighting the critical role of phonon-assisted relaxation in mid-IR emission. Finally, the potential of other Dy³⁺-doped fluorite-type hosts, such as SrF₂ and BaF₂, is evaluated.

1. Introduction

Coherent light sources addressing the 3–4 μm mid-infrared spectral range are of strong interest because their emission coincides with fundamental vibrational absorption bands of many molecules, including hydrocarbons, water, and organic functional groups [1]. This makes them relevant for spectroscopy, remote sensing, and environmental monitoring, where strong molecular signatures enable sensitive detection [2–4]. This wavelength region is also well suited for medical applications, as it overlaps with absorption peaks in biological tissues, allowing precise and minimally invasive incisions. Moreover, 3–4 μm sources are valuable for nonlinear optics and mid-IR frequency conversion, where they serve as pump or seed wavelengths for extending laser technologies deeper into the infrared [5].

Direct generation of 3 μm mid-infrared radiation from rare-earth-ion (RE³⁺) doped materials is feasible with Er³⁺, Ho³⁺ and Dy³⁺ ions, out of

which crystalline and fiber Er-lasers operating on the ⁴I_{11/2} → ⁴I_{13/2} transition are the most widely exploited ones due to efficient pumping at 0.98 μm enabled by InGaAs diode lasers [6–8]. Dysprosium ions (Dy³⁺) possessing an electronic configuration of [Xe]4f⁹ (with ⁶H_{15/2} being the ground state) has attracted a growing attention for laser action in this spectral range as their ultrabroadband emission originating from the quasi-3-level ⁶H_{13/2} → ⁶H_{15/2} transition extends far beyond 3 μm, *i.e.*, it is well detuned from the structured water vapor absorption in air at 2.8 μm, a property being essential for broadband wavelength tuning and, potentially, achieving femtosecond pulses from mode-locked oscillators. The drawbacks of Dy³⁺ include a significant non-radiative path from the ⁶H_{13/2} manifold reducing the luminescence quantum yield at room temperature (even in fluoride matrices with reasonably low phonon energies), as well as challenging pumping: to maintain low quantum defect, one needs to address non-conventional pump wavelengths in the near-infrared such as 1.1, 1.3, 1.7 or 2.8 μm [8]. Note that

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Dy³⁺ also offers other emissions in the mid-IR, Fig. 1, in particular, at 4.4 μm (the ⁶H_{11/2} → ⁶H_{13/2} transition) [9]. However, potential laser operation is challenged by the bottleneck effect: the lifetime of the terminal laser level is much longer than that of the emitting state, in part due to the intrinsically higher probability of multiphonon non-radiative relaxation from the ⁶H_{11/2} manifold [10].

Continuous-wave laser action around 3 μm (on the ⁶H_{13/2} → ⁶H_{15/2} transition) from Dy³⁺-doped materials at room temperature has been only reported from fluoride glass (ZrF₄- or InF₃-based) fibers. Due to the adaptability of fiber geometry to non-conventional pumping, multiple excitation wavelengths have been exploited so far. Woodward et al. reported on a Dy:ZBLAN fiber laser with in-band (resonant) pumping at 2.83 μm delivering Watt-level output power at 3.15 μm with a high slope efficiency (73%) approaching the Stokes limit [11]. Pumping at 1.7 μm and 1.1 μm (into the ⁶H_{11/2} and ⁶F_{9/2}+⁶H_{7/2} manifolds, respectively) using a Raman-shifted fiber laser and an Yb-fiber source resulted in lower slope efficiencies of 21% and 18%, respectively, achieving a few hundred of mW output powers [12,13]. Regarding single crystals, Johnson and Guggenheim achieved laser emission at 3.02 μm from Dy:BaY₂F₈ under liquid nitrogen temperature [14].

To overcome these limitations, the Er³⁺,Dy³⁺ codoping scheme was proposed as it can simultaneously unlock access to conventional pumping at 0.98 μm, and broaden the overall gain bandwidth around 3 μm by combining the spectral emission profiles from both active ions. The excitation of Er³⁺ ions into the ⁴I_{11/2} state is followed by a two-step phonon-assisted energy-transfer (ET) to Dy³⁺ ions, as shown in Fig. 1. Wang et al. reported on an Er,Dy:ZBLAN fiber laser broadly tunable from 2.71 to 3.37 μm [15].

Regarding the second appealing mid-IR laser transition of Dy³⁺ around 4.4 μm, very little information can be found in the literature. Barnes and Allen reported on a room temperature Dy:LiYF₄ laser at 4.34 μm operating in the free-running regime [9]. Pulsed laser emission from chalcogenide, i.e., Dy:CaGa₂S₄ and Dy:PbGa₂S₄ [10,16], and chloride, i.e., Dy:KPB₂Cl₅, crystals has also been achieved. Schlosser et al. demonstrated a 1.7-μm diode-pumped Dy:KPB₂Cl₅ laser delivering 1.1 mJ pulses at ~4.4 μm with 12% slope efficiency [17].

Since mid-IR transitions of RE³⁺ ions are strongly affected by non-radiative multiphonon relaxation, host matrices with low phonon energies are desirable. Fluoride crystals, compared with more technologically challenging systems such as chlorides and chalcogenides, represent an attractive compromise between low phonon energy and

high optical quality. In particular, fluorite-type alkaline-earth fluorides MF₂ (M = Ca, Sr, Ba, or their mixture) [18,19] are well-established laser host materials due to their high thermal conductivity, low phonon energies, broad optical transparency, and high RE solubility [20,21] retaining the cubic, fluorite-type structure (M_{1-x}RE_xF_{2+x}, sp. gr. *Fm* $\bar{3}m$) [22]. Moreover, at moderate to high doping levels, RE³⁺ ions tend to form clusters [22], leading to strong inhomogeneous broadening of absorption and emission (gain) spectra [23]. As a result, the spectral properties of rare-earth-doped MF₂ crystals more closely resemble those of fluoride glasses than those of ordered crystals such as LiYF₄, while still maintaining relatively high thermal conductivity. The growth of RE³⁺-doped MF₂ crystals is well developed using the Bridgman–Stockbarger and Czochralski methods; however, precise control of the growth atmosphere is required to suppress oxygen impurities that can degrade luminescent performance.

Er³⁺ and Ho³⁺ doped MF₂-type crystals have been recognized as effective gain media for lasers operating around the 3 μm spectral region [18,24,25]. In contrast, studies on the mid-infrared emission properties of Dy³⁺-doped fluorite-type crystals – particularly CaF₂ – covering the 3–4.4 μm range remain scarce. Therefore, in the present work, we systematically investigate the spectroscopic properties of Dy³⁺ ions in calcium fluoride crystals for mid-infrared laser applications, focusing on their stimulated-emission properties at 3 μm and 4.4 μm, radiative and non-radiative transition probabilities, and the energy-transfer mechanisms in the Er³⁺–Dy³⁺ co-doped system.

2. Crystal growth

Single crystals of calcium fluoride (CaF₂), singly doped with Dy³⁺ and codoped with Er³⁺,Dy³⁺ ions, were grown using the Bridgman–Stockbarger technique. The starting powders were high-purity (4 N, *Sigma-Aldrich*) CaF₂ and dysprosium fluoride (DyF₃), the latter of which was obtained by fluorinating dysprosium oxide (Dy₂O₃, 4 N, *Alfa Aesar*) with excess of ammonium hydrogen bifluoride (NH₄HF₂) at 250 °C. For the singly Dy³⁺-doped crystal, the starting doping level was 0.5 at.% with respect to Ca²⁺ cations (Ca_{1-x}Dy_xF_{2+x}) and for the three codoped crystals, the Er concentration was maintained at 3 at.%, while Dy content was 0.1, 0.3 or 0.5 at.%. To avoid the formation of unwanted oxygen-assisted centers and oxyfluoride phases which can degrade the optical quality and spectroscopic properties, the growth chamber was first sealed to vacuum (<10⁻⁵ mbar) and refilled with a mixture of Ar + CF₄ gases. The precursors were well mixed and placed in the graphite crucible which was then heated at a temperature slightly above the CaF₂ melting point (1418 °C) for 2 h to ensure uniform melting. The crystal growth was enabled by a vertical translation of the graphite crucibles (diameter: Ø8 mm, height: 40 mm) through the hot zone with a temperature gradient of 30 – 40 °C/cm at a rate of 1 – 3 mm/h. Once the growth was completed, the crystals were gradually cooled to room temperature within 72 h. No annealing was applied.

The as-grown crystals had a cylindrical shape (Ø8 mm, length: 35–40 mm) and they were transparent, without cellular substructure, free of defects and inclusions, Fig. 2. The singly Dy³⁺-doped crystal was colorless, and the codoped ones had a rose coloration due to Er³⁺ absorption. For spectroscopic studies, we cut cylindrical samples from the central part of the boule having a thickness of 5 mm. The sample was then polished to laser-grade quality on both sides, see Fig. 2(b). The segregation coefficient of Dy³⁺ dopant ions was determined by inductively coupled plasma mass spectrometry (ICP–MS) to be close to unity (*K*_{Dy} ≈ 1). The axial distribution of dopant along the cylindrical part of the boule was subsequently assessed by absorption spectroscopy and was found to be uniform, with variations not exceeding 0.05 at.% across 20 mm length in the singly doped crystal.

Crystals adopting the fluorite structure exhibit small enthalpy and entropy changes upon melting. During solidification, their growth is often characterized by the formation of dissipative structures, such as

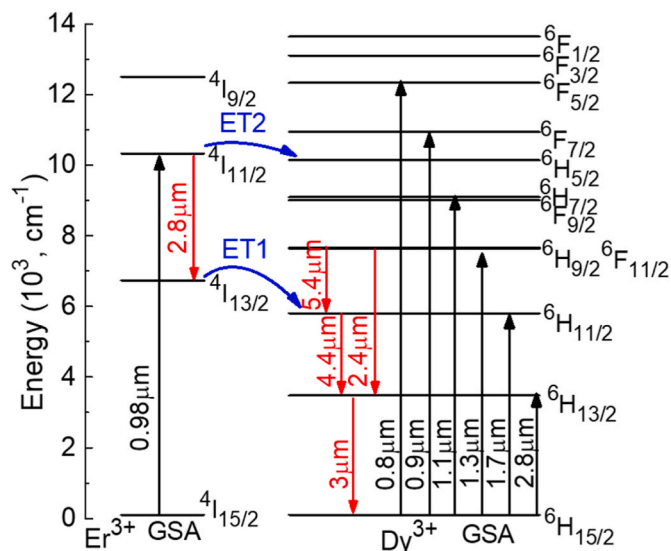


Fig. 1. A partial energy-level scheme of Dy³⁺ and Er³⁺ ions showing pump and laser transitions, GSA – ground-state absorption, ET – phonon-assisted non-radiative energy-transfer.

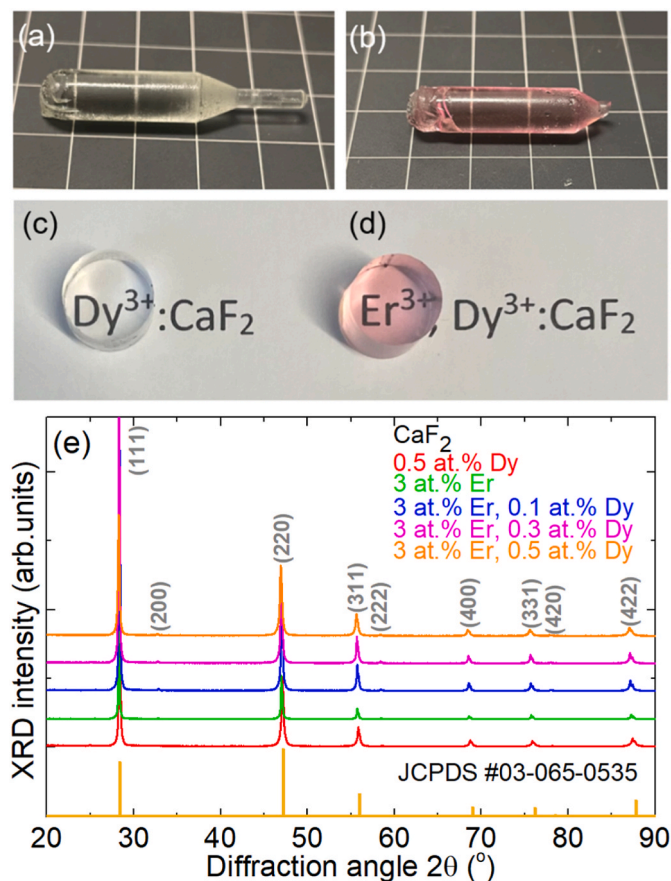


Fig. 2. The photographs of (a,b) as-grown crystal boules and (c,d) laser-grade polished samples of (a,c) singly doped 0.5 at.% Dy³⁺:CaF₂ and (b,d) codoped 3 at.% Er³⁺, 0.1 at.% Dy³⁺:CaF₂. (e) X-ray powder diffraction patterns of the studied crystals, the theoretical pattern for CaF₂ is given (JCPDS card #03-065-0535), (hkl) – Miller's indices.

cellular and dendritic morphologies, arising from crystallization of the melt [26]. Optical inhomogeneities are typically attributed to periodic refractive-index variations associated with compositional fluctuations. The instability of the planar solid–liquid interface can be suppressed by employing appropriate growth conditions, specifically a high temperature gradient and a low growth rate, thereby enabling the formation of homogeneous single crystals, as achieved in the present work.

The phase purity (sp. gr. $Fm\bar{3}m - O_h^5$, No. 225) of the grown crystals has been confirmed using X-ray powder diffraction (XRD) employing a Rigaku SmartLab diffractometer (step size: 0.005°, step time: 0.06 s, 2θ range: 20°–95°), Fig. 2(e).

3. Experimental

The thermal conductivity $\kappa(T)$ of a 1 at.% Dy:CaF₂ crystal in the temperature range of 50–300 K was measured by the absolute steady-state longitudinal heat-flow method. The sample had a cylindrical shape with a diameter of 9.6 mm and a length of 37 mm (the analyzed distance corresponds to the 19 mm separation between the thermocouple contact positions). To create a flat (planar) heat front and a uniform distribution of the heat flux along the sample, a resistive heater was attached to one of end faces. The temperature difference along the crystal, induced by the heater, did not exceed 1 K and was monitored using a chromel–(copper + iron) thermocouple. The thermal conductivity coefficient $\kappa(T)$ was calculated using Fourier law, with a measurement error of ±5%. More details can be found elsewhere [27].

The Raman spectra were collected using a confocal Raman

microscope (inVia Qontor, Renishaw), equipped with a ×50 Leica objective and an Ar⁺ ion laser ($\lambda_{exc} = 488 \text{ nm}, 514 \text{ nm}$).

The absorption spectra were measured using a Varian CARY 5000 spectrophotometer with a spectral resolution of 0.4 nm to 1.75 nm, depending on the spectral range. The spectra of Dy³⁺ luminescence around 3 μm were measured using an optical spectrum analyzer (OSA, Yokogawa AQ6377E) with a spectral resolution of 2 nm. A 1320 nm diode laser served as excitation source. Emission was collected through an uncoated CaF₂ lens and coupled directly into the OSA purged with N₂ gas. The setup was calibrated using a SiC globar lamp (ARCLIGHT-MIR, Arcoptix).

The luminescence decay from the ⁴I_{13/2} and ⁴I_{11/2} Er³⁺ manifolds was studied using a nanosecond optical parametric oscillator (Horizon, Continuum) as excitation source and detected using a set-up consisting of a monochromator (Oriol 77200), InGaAs detectors (Hamamatsu), a pre-amplifier (DHPCA-100, FEMTO) and an 8 GHz digital oscilloscope (DSA70804B, Tektronix). The decay from the ⁶H_{13/2} Dy³⁺ state was measured using 808 nm laser diode as excitation source and detected using a system comprising a monochromator (Oriol 74086), a lock-in amplifier (SR810 DSP, Stanford Research Systems), and a liquid nitrogen-cooled InSb detector (J10D series, Judson Infrared) and the same oscilloscope. The same setup was also employed for measuring the exponential sideband of Dy³⁺ emission between 3.4 and 4 μm.

4. Results and discussions

4.1. Thermal conductivity

The temperature dependence of thermal conductivity of 1 at.% Dy:CaF₂ is illustrated in Fig. 3. Dy³⁺ doping reduces the value of κ as compared to the nominally pure (undoped) crystal, namely from 10.3 Wm⁻¹K⁻¹ [28] to 7.0 Wm⁻¹K⁻¹ at 300 K, while it remains reasonably high for potential laser applications. One can observe a low-temperature maximum at ~120 K on the $\kappa(T)$ curve. According to the previous studies, for CaF₂ crystals doped with rare-earth ions of the yttrium subgroup (including Dy³⁺), the dominating type of the formed ion clusters is the large octacubic cluster (RE₆F₃₇)¹⁹⁻ [29–31]. These structures act as efficient centers of phonon scattering resulting in the observed characteristic shape of the temperature dependence of the thermal conductivity. For comparison, the same figure includes reference data for 3 at.% Er³⁺:CaF₂, which exhibits a lower thermal conductivity at room temperature (4.3 Wm⁻¹K⁻¹), suggesting that codoped crystals are likely to experience pronounced thermal effects.

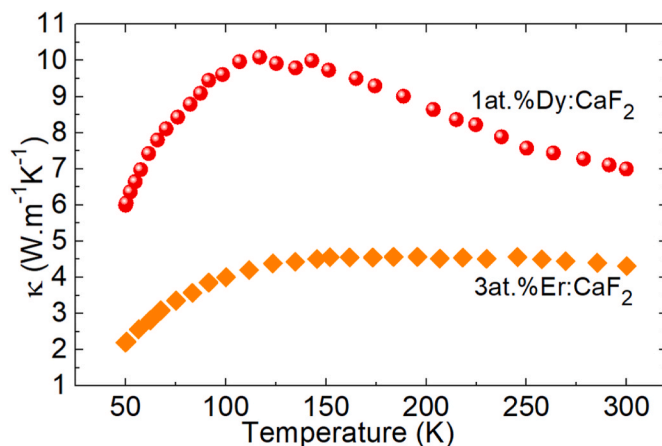


Fig. 3. Temperature dependence of thermal conductivity for 1 at.% Dy³⁺:CaF₂ and 3 at.% Er³⁺:CaF₂ crystals.

4.2. Optical absorption

For the singly Dy³⁺-doped CaF₂ crystal, the absorption bands in the transparency range are related to $4f - 4f$ electronic transitions of Dy³⁺ ion from its ground state, ⁶H_{15/2}. The absorption cross-sections, $\sigma_{abs} = \alpha_{abs}/N_{Dy}$, were calculated from the absorption coefficient α_{abs} and ion density $N_{Dy} = 1.23 \times 10^{20}$ at/cm³, calculated using the segregation coefficient of Dy³⁺ ions, $K_{Dy} \approx 1$. The σ_{abs} spectra are plotted in Fig. 4(a–f). The electronic transitions of Dy³⁺ ions were assigned according to the energy-level scheme of the free ion reported by Carnall et al. [32]. The absorption bands are broad and almost structureless (they resemble those in Dy³⁺-doped fluoride glasses, revealing a “glassy-like” behavior) due to the strong inhomogeneous spectral line broadening. The latter originates from the rare-earth ion clustering in CaF₂.

There exist various pumping schemes for mid-infrared Dy³⁺ lasers, addressing the absorption transitions at 0.8 μ m (⁶H_{15/2} → ⁶H_{5/2}), 0.9 μ m (⁶H_{15/2} → ⁶H_{7/2}), 1.1 μ m (⁶H_{15/2} → ⁶H_{7/2}+⁶F_{9/2}), 1.3 μ m (⁶H_{15/2} → ⁶H_{9/2}+⁶F_{11/2}), 1.7 μ m (⁶H_{15/2} → ⁶H_{11/2}), as well as 2.8 μ m (⁶H_{15/2} → ⁶H_{13/2}) [11–13,33,34]. When targeting laser emission at 3 μ m, resonant (in-band) pumping directly into the upper laser level (⁶H_{13/2}) is expected to lead to the lowest quantum defect and highest slope efficiency,

however, it requires customized pump sources, such as Er-fiber lasers [11]. For the ⁶H_{15/2} → ⁶H_{13/2} Dy³⁺ transition in CaF₂, the peak $\sigma_{abs} = 0.35 \times 10^{-20}$ cm² at 2834 nm (absorption bandwidth (FWHM): 144 nm). Progressively reducing the pump wavelength allows one to access more conventional pump sources at the expense of reduced Stokes efficiency and severe thermal effects. For the ⁶H_{15/2} → ⁶H_{11/2} Dy³⁺ transition addressed by Raman-shifted Er-fiber lasers [12], the peak $\sigma_{abs} = 0.16 \times 10^{-20}$ cm² at 1692 nm (absorption bandwidth: ~138 nm). Note that this pump scheme will correspond to the resonant excitation of the 4.4- μ m laser transition of Dy³⁺ (⁶H_{11/2} → ⁶H_{13/2}).

Dysprosium is a polyvalent ion, and Dy²⁺ can exist in fluorite type crystals similarly to some other divalent-metal rare-earths, notably, Yb²⁺, Tm²⁺, Eu²⁺ or Sm²⁺ [35]. Dy²⁺ ions exhibit characteristic absorption in the visible which is not observed in the present work. Dy²⁺:CaF₂ is one of the earliest laser materials based on both ceramic and single-crystal architectures [36–39].

4.3. Judd-Olfelt theory

The intensities of $4f - 4f$ transitions of Dy³⁺ ions in the 0.5 at.% Dy:CaF₂ crystal were calculated based on the measured absorption spectrum within the modified Judd-Olfelt (mJ-O) theory accounting for weak

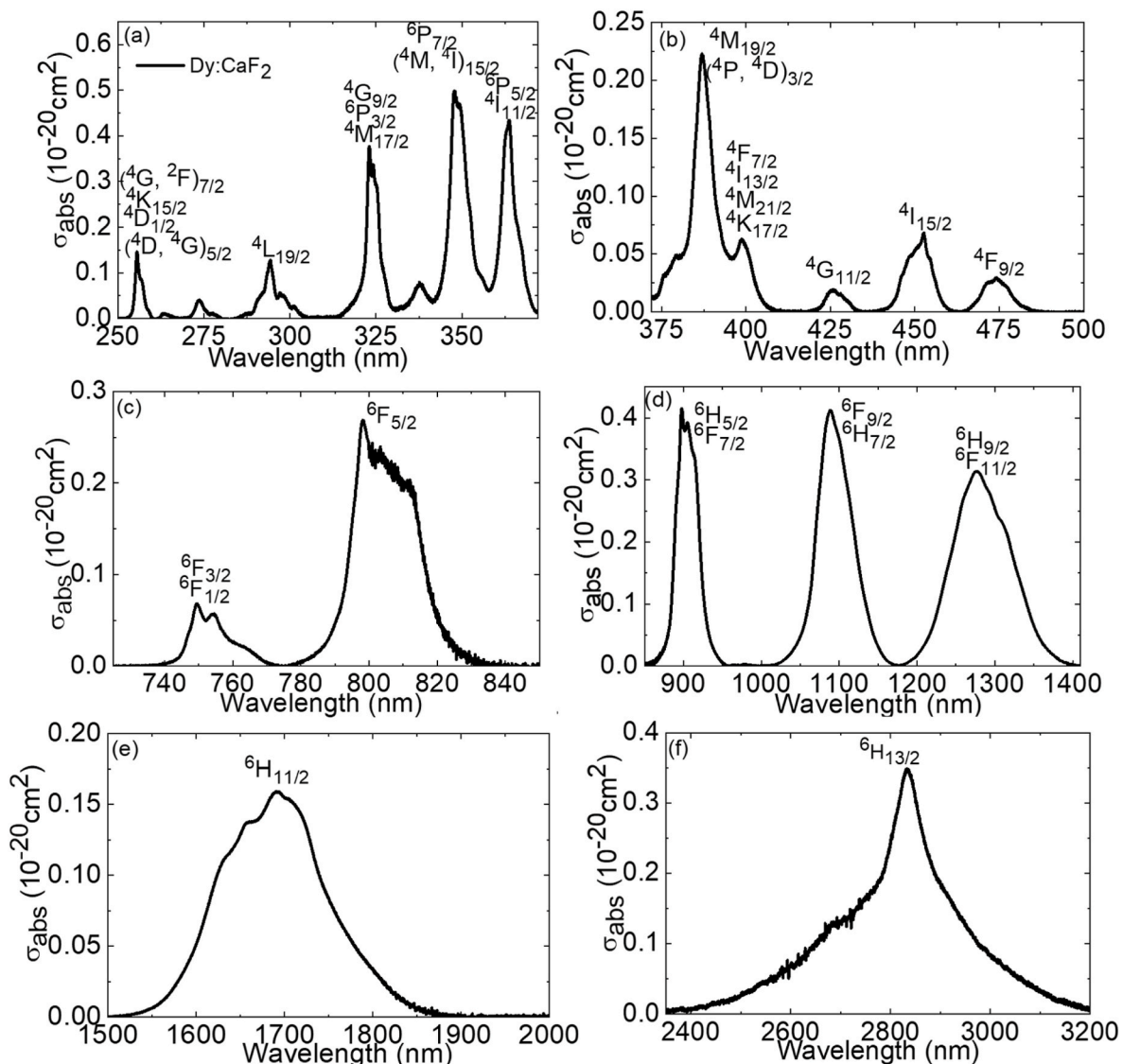


Fig. 4. (a–f) Absorption cross-section (σ_{abs}) spectra of 0.5 at.% Dy³⁺:CaF₂ crystal.

configuration interaction with the excited configuration of the opposite parity, $4f^{n-1}5d^1$. Within this parametrization scheme, the electric dipole (ED) line strengths of $4f-4f$ transitions are given by Refs. [40,41]:

$$S_{ff}^{ED}(JJ) = \sum_{k=2,4,6} U^{(k)} \tilde{\Omega}_k, \quad (1a)$$

$$\tilde{\Omega}_k = \Omega_k \left[1 + 2\alpha(E_J - E_{J'} - 2E_f^0) \right], \quad (1b)$$

where $U^{(k)}$ are the reduced squared matrix elements, and the intensity parameters $\tilde{\Omega}_k$ vary linearly with the energies (E_J and $E_{J'}$) of the two multiplets participating in the optical transition $J \rightarrow J'$, E_f^0 represents the mean energy of the $4f^n$ ground-state configuration and the value of the α parameter is determined by two equivalent mechanisms: covalence effects and excited configurations of opposite parity. Consequently, within the mJ-O theory, there are four free (fitting) parameters, Ω_2 , Ω_4 , Ω_6 , and α .

The experimental, f_{exp} , and calculated, f_{calc} , oscillator strengths for Dy^{3+} transitions observed in absorption are listed in Table 1. The contributions from electric dipole (ED) and magnetic dipole (MD, for $\Delta J = 0, \pm 1$) transitions are included. The latter were calculated separately using the Russell-Saunders approximation on wave functions of Dy^{3+} ions under the assumption of a free-ion. The set of $U^{(k)}$ was calculated using the free-ion parameters reported by Carnall et al. [32]. The refractive index of CaF_2 was calculated based on the dispersion relation from Ref. [42]. Γ refers to the absorption coefficient integrated over a band corresponding to a particular transition or a group of undistinguishable (spectrally overlapping) transitions. The use of the mJ-O theory allows to improve the agreement between $\langle f_{exp} \rangle$ and $\langle f_{calc} \rangle$, reducing the root mean square (r.m.s.) deviation from 0.521 for the standard J-O theory to 0.280 for the mJ-O one.

The following intensity parameters were obtained: $\Omega_2 = 0.708$, $\Omega_4 = 4.179$, $\Omega_6 = 3.666 [10^{-20} \text{ cm}^2]$ and $\alpha = 0.91 \times 10^{-4} \text{ cm}$ ($\Delta = 0.55 \times 10^4 \text{ cm}^{-1}$) for the mJ-O theory and $\Omega_2 = 0.067$, $\Omega_4 = 4.440$, $\Omega_6 = 2.612 [10^{-20} \text{ cm}^2]$ for the standard parametrization scheme. Gao et al. previously investigated Dy^{3+} -doped and Dy^{3+}, Y^{3+} -codoped CaF_2

Table 1

Intensities^a of transitions in absorption of dysprosium (III) ions in the 0.5 at.% $Dy^{3+}:CaF_2$ crystal.

${}^6H_{15/2} \rightarrow$	$\langle E \rangle$,	Γ ,	$f^c, 10^{-6}$		
	cm^{-1}	$\text{cm}^{-1} \cdot \text{nm}$	f_{exp}	$f_{calc}^{(J-O)}$	$f_{calc}^{(mJ-O)}$
${}^6H_{13/2}$	3537.4	129.720	1.503	1.430 ^{ED+} 0.330 ^{MD}	1.144 ^{ED+} 0.330 ^{MD}
${}^6H_{11/2}$	6007.9	34.464	1.117	0.950 ^{ED}	0.868 ^{ED}
${}^6H_{9/2} + {}^6F_{11/2}$	7750.8	45.557	2.570	3.245 ^{ED}	2.598 ^{ED}
${}^6F_{9/2} + {}^6H_{7/2}$	9125.9	34.727	2.678	3.534 ^{ED}	2.874 ^{ED}
${}^6H_{5/2} + {}^6F_{7/2}$	11034	20.312	2.300	2.378 ^{ED}	2.301 ^{ED}
${}^6F_{5/2}$	12416	8.316	1.189	0.995 ^{ED}	1.078 ^{ED}
${}^6F_{3/2} + {}^6F_{1/2}$	13239	1.857	0.300	0.189 ^{ED}	0.209 ^{ED}
${}^4F_{9/2}$	21092	0.488	0.201	0.178 ^{ED}	0.216 ^{ED}
${}^4I_{15/2}$	22202	0.926	0.421	0.426 ^{ED+} 0.086 ^{MD}	0.546 ^{ED+} 0.086 ^{MD}
${}^4G_{11/2}$	23409	0.246	0.125	0.132 ^{ED}	0.122 ^{ED}
${}^4F_{7/2} + {}^4I_{13/2}$ $2 + {}^4M_{21/2}$ $2 + {}^4K_{17/2}$	25798	3.910	2.406	2.269 ^{ED+} 0.007 ^{MD}	2.830 ^{ED+} 0.007 ^{MD}
${}^6P_{5/2} + {}^4I_{11/2}$ $2 + {}^6P_{7/2}$ $2 + {}^4M_{15/2} +$ ${}^4D_{5/2} + {}^4I_{9/2}$	28637	9.850	7.310	6.749 ^{ED+} 0.010 ^{MD}	7.216 ^{ED+} 0.010 ^{MD}
${}^4G_{9/2} + {}^6P_{3/2}$ $2 + {}^4M_{17/2}$	30771	2.068	1.822	0.911 ^{ED}	1.408 ^{ED}
<i>r.m.s deviation</i>				0.521	0.280

^a $\langle E \rangle$ – energy of the “center of gravity” of the absorption band, Γ – integrated absorption coefficient; f_{exp} and f_{calc}^c – experimental and calculated absorption oscillator strengths, respectively, ED and MD – electric-dipole and magnetic-dipole contributions, respectively.

with respect to their visible emission properties. For the Dy^{3+} -doped crystal, they reported the intensity parameters of $\Omega_2 = 0.899$, $\Omega_4 = 0.796$, $\Omega_6 = 2.429 [10^{-20} \text{ cm}^2]$ using the standard J-O analysis [43].

The probabilities of spontaneous radiative transitions of Dy^{3+} ions were calculated using the mJ-O model. The parameters relevant for laser operation in the mid-infrared range are listed in Table 2, namely, the probabilities of spontaneous radiative transitions $A_{calc}^{\Sigma}(JJ)$, the radiative lifetimes τ_{rad} of the ${}^6H_{13/2}$, ${}^6H_{11/2}$ and ${}^6H_{9/2}$ states, and the luminescence branching ratios $\beta_{JJ'}$ for the ${}^6H_{9/2} \rightarrow {}^6H_{13/2}$, ${}^6H_{11/2} \rightarrow {}^6H_{13/2}$, and ${}^6H_{9/2} \rightarrow {}^6H_{11/2}$ transitions giving rise to emission at 2.4 μm , 4 μm and 5.7 μm , respectively.

In particular, the radiative lifetime $\tau_{rad}({}^6H_{13/2})$ is 46.67 ms, in good agreement with previous values obtained for a $Dy^{3+}, Y^{3+}:CaF_2$ crystal using standard J-O theory (51.15 ms) [43], as well as with the estimate reported by Fleischman et al. for $Dy^{3+}:BaF_2$ (45.9 ms), derived from a comparison between the reciprocity method and the Fichtbauer-Ladenburg approach for stimulated-emission cross-section evaluation [44].

4.4. Stimulated-emission and gain cross-sections at $\sim 3 \mu\text{m}$

The stimulated-emission (SE) cross-sections, σ_{SE} , for the ${}^6H_{13/2} \rightarrow {}^6H_{15/2}$ Dy^{3+} transition at $\sim 3 \mu\text{m}$ were calculated using the Fichtbauer-Ladenburg (F-L) equation [45]:

$$\sigma_{SE}(\lambda) = \frac{\lambda^5}{8\pi n^2 \tau_{rad} c} \times \frac{\beta(JJ') I(\lambda)}{\int \lambda I(\lambda) d\lambda}, \quad (2)$$

where λ is the wavelength of light, n is the refractive index of the crystal at the mean emission wavelength, c is the speed of light, τ_{rad} is the radiative lifetime of the emitting state (${}^6H_{13/2}$), $\beta_{JJ'} = 1$, and $I(\lambda)$ is the measured spectral profile of luminescence for the considered transition calibrated for the apparatus function.

Fig. 5(a) depicts the absorption and SE cross-section spectra of singly Dy^{3+} -doped CaF_2 crystal in the spectral range around 3 μm corresponding to ${}^6H_{15/2} \leftrightarrow {}^6H_{13/2}$ transition. A significant spectral broadening is also observed for the emission profile, resulting in a broad emission bandwidth (FWHM) of 350 nm. Remarkably, the emission extends far beyond 3 μm avoiding the structured water vapor absorption in the atmosphere. The peak σ_{SE} value reaches $0.25 \times 10^{-20} \text{ cm}^2$ at 2947 nm.

3- μm Dy lasers operate on a quasi-3-level scheme due to the inherent reabsorption from the ground-state. Thus, gain cross-sections, $\sigma_{gain} = \sigma_{SE} \times \beta - \sigma_{abs} \times (1 - \beta)$, are calculated to assess the gain bandwidth, the expected laser wavelength and the tuning range, where $\beta = N_2({}^6H_{13/2})/N_{Dy}$ is the population inversion ratio. Fig. 5(b) displays the spectral gain profiles for $Dy:CaF_2$ around 3 μm . On increasing β from 0.2 to 0.45, the maximum in the gain spectra experiences a blue shift from 3284 nm to 3030 nm. At an intermediate β of 0.35, the gain bandwidth (FWHM) is as broad as 338 nm. The broadband gain properties of $Dy^{3+}:CaF_2$ crystal around 3 μm indicate its potential for generation and amplification of extremely short pulses – down to femtosecond regime and the tunability at these wavelengths.

Table 2

Selected probabilities of spontaneous radiative transitions of Dy^{3+} in CaF_2 crystal.

Transition	$\langle \lambda_{em} \rangle$, nm	$A_{calc}^{\Sigma}(JJ)$, s^{-1}	$\beta_{JJ'}$, %	τ_{rad} , ms
${}^6H_{13/2} \rightarrow {}^6H_{15/2}$	2826.9	15.57 ^{ED} +6.33 ^{MD}	100	45.67
${}^6H_{11/2} \rightarrow {}^6H_{13/2}$	4047.8	3.89 ^{ED} +3.69 ^{MD}	11.9	15.63
${}^6H_{9/2} \rightarrow {}^6H_{11/2}$	5737.6	3.11 ^{ED} +1.45 ^{MD}	1.3	2.81
$\rightarrow {}^6H_{13/2}$	2373.4	62.39 ^{ED} +0.68 ^{MD}	17.7	

^a $\langle \lambda_{em} \rangle$ – mean wavelength of the emission band, $\beta_{JJ'}$ – luminescence branching ratio, τ_{rad} – radiative lifetime of the excited state, ED and MD – electric-dipole and magnetic-dipole, respectively.

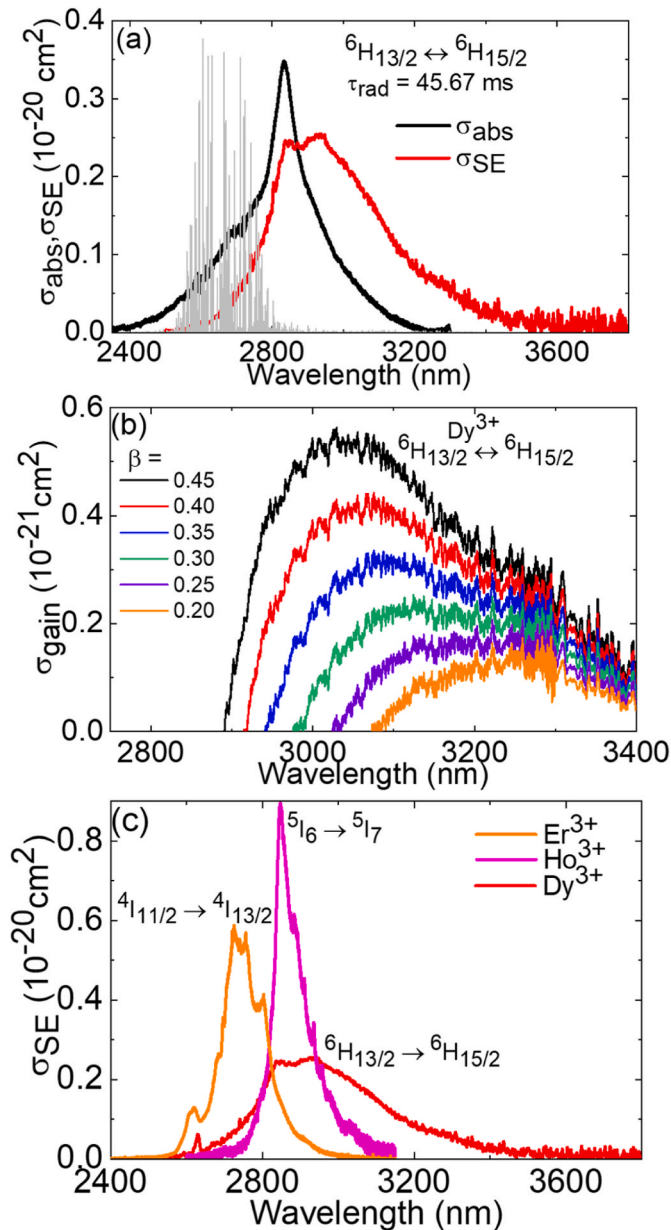


Fig. 5. (a) Absorption, σ_{abs} , and stimulated emission (SE), σ_{SE} , cross-sections for the ${}^6H_{15/2} \leftrightarrow {}^6H_{13/2}$ transitions of Dy^{3+} ions in CaF_2 crystal, corrected for the structured water vapor absorption in air (in grey, arb. units, according to the HITRAN database); (b) Gain cross-section spectra, $\sigma_{gain} = \sigma_{SE} \times \beta - \sigma_{abs} \times (1 - \beta)$, around 3 μm plotted for different population inversion ratios (β); (c) Comparison of σ_{SE} spectra around 3 μm for Er^{3+} , Ho^{3+} and Dy^{3+} ions in CaF_2 .

It is worth comparing the SE cross-section spectra of three rare-earth ions exhibiting luminescence around 3 μm , notably Er^{3+} , Ho^{3+} and Dy^{3+} in CaF_2 crystals, see Fig. 5(c). Among these three ions, Dy^{3+} offers the broadest emission bandwidth covering the range far beyond 3 μm , i.e., being well detuned from the structured water vapor absorption in air, a property being essential for achieving femtosecond pulses from mode-locked oscillators and unlike for Er^{3+} and Ho^{3+} , the ${}^6H_{13/2} \rightarrow {}^6H_{15/2}$ Dy^{3+} transition is not of self-terminating nature. However, σ_{SE} values for Dy^{3+} ions around 3 μm are about two times lower than those for Er^{3+} and Ho^{3+} ones. This is due to a combination of a relatively long radiative lifetime of the ${}^6H_{13/2}$ manifold of Dy^{3+} , as well as the large ground-state splitting of this ion, which results in a broadened emission band.

With the goal of acquiring the broader gain bandwidth around 3 μm , it is obvious that a combination of Er^{3+} and Dy^{3+} will be the prospect for

the broadly tunable laser potentially covering the wavelength range spanning from 2.7 μm to 3.3 μm as demonstrated earlier for $Dy:ZBLAN$ glass fibers [15].

4.5. Emission properties of Er^{3+}, Dy^{3+} -codoped crystals

Codoping of CaF_2 crystals with both Er^{3+} and Dy^{3+} ions serves two aims: i) to combine the spectral gain profiles resulting in broadband emission, and ii) to resolve the main difficulty in exciting mid-infrared Dy^{3+} emission, namely, the relatively weak absorption of this ion and its non-conventional pump wavelengths. The donor (D) Er^{3+} ions can be efficiently pumped at 0.96 μm into the ${}^4I_{11/2}$ manifold, e.g., by InGaAs diode lasers or Yb-fiber lasers, and there exist two phonon-assisted non-radiative energy-transfer (ET) processes, ${}^4I_{11/2}(Er^{3+}) \rightarrow {}^6H_{5/2}(Dy^{3+})$ and ${}^4I_{13/2}(Er^{3+}) \rightarrow {}^6H_{11/2}(Dy^{3+})$, enabling acceptor (A) Dy^{3+} emission around 3 μm and 4.4 μm .

Crystals codoped with 3 at.% Er^{3+} and a various content of Dy^{3+} (ranging from 0.1 to 0.5 at.%) were studied. By exciting the luminescence using an InGaAs laser diode at 960 nm (to the ${}^4I_{11/2}$ Er^{3+} state), broadband emission around 3 μm originating from both ions was observed. On increasing the Dy content, the emission spectrum progressively broadened extending far beyond 3 μm , and the contribution of this long-wave emission gradually increased, Fig. 6(a). The broadest emission profile was observed for the 3 at.% Er^{3+} , 0.5 at.% Dy^{3+} codoped CaF_2 crystal, exhibiting a bandwidth (FWHM) of 310 nm. By studying the corresponding singly Er^{3+} and Dy^{3+} -doped crystals, the spectrum was deconvoluted and the emission at shorter wavelengths with the most intense peak at 2.8 μm was assigned to Er^{3+} ions (${}^4I_{11/2} \rightarrow {}^4I_{13/2}$), while that at longer wavelengths belongs to Dy^{3+} ions (${}^6H_{13/2} \rightarrow {}^6H_{15/2}$).

We have also studied the long-wavelength sideband of the ${}^6H_{13/2} \rightarrow {}^6H_{15/2}$ emission band of Dy^{3+} ions, by sacrificing the spectral resolution to gain higher sensitivity to this weak emission. This emission is due to a phonon-terminated (Stokes) process, and its spans up to at least 3.9 μm , see Fig. 6(c). Note that both Raman (321 cm^{-1}) and IR (466 cm^{-1}) active phonons may participate in this emission. The longest wavelength of a purely electronic transition for Dy^{3+} ions in C_{4v} symmetry sites in CaF_2 is 3.32 μm (3622 cm^{-1} (${}^6H_{13/2}, \Delta_7$) \rightarrow 616 cm^{-1} (${}^6H_{15/2}, \Delta_7$), Δ_i – irreducible representation) [46].

4.6. Luminescence dynamics

First, the luminescence decay from the ${}^6H_{13/2}$ Dy^{3+} manifold was studied. For the crystal singly doped with 0.5 at.% Dy^{3+} , the mean luminescence lifetime $\langle \tau_{lum} \rangle$ was measured to be 0.41 ms (the rate of non-radiative relaxation: $W_{NR} = 2.42 \times 10^3 \text{ s}^{-1}$). Monitoring the luminescence lifetime of the same manifold for the 3 at.% Er^{3+} , x at.% Dy^{3+} codoped CaF_2 crystals, a close value was found for $x = 0.5$ ($\langle \tau_{lum} \rangle = 0.52 \text{ ms}$), while a significant increase of the lifetime was observed for crystals with $x = 0.3$ (0.81 ms) and $x = 0.1$ (2.04 ms), see Fig. 7(a). Still, these values are much shorter than the radiative lifetime ($\tau_{rad} = 45.7 \text{ ms}$) due to the non-negligible multiphonon non-radiative relaxation.

The luminescence lifetimes of the ${}^4I_{11/2}$ and ${}^4I_{13/2}$ manifolds of donor Er^{3+} ions in codoped crystals (denoted as $\tau_{D,A}$) were then measured as a function of Dy^{3+} (acceptor) doping level under resonant excitation and compared with those (denoted as τ_D) for a singly 3 at.% Er^{3+} -doped CaF_2 crystal, see Fig. 7(b and c). For the singly-doped crystal, one finds $\tau_D = 3.32 \text{ ms}$ (${}^4I_{13/2}$) and 5.72 ms (${}^4I_{11/2}$). On increasing the Dy^{3+} doping level x, the $\tau_{D,A}$ lifetimes progressively decrease reaching 0.20 ms (${}^4I_{13/2}$) and 0.44 ms (${}^4I_{11/2}$) for $x = 0.5$, indicating an efficient energy transfer $Er^{3+}(D) \rightarrow Dy^{3+}(A)$ promoted by the close proximity of rare-earth ions forming clusters in CaF_2 .

The efficiency of $Er^{3+} \rightarrow Dy^{3+}$ energy transfer can be estimated from the measured lifetimes as $\eta_{ET} = 1 - \tau_{D,A}/\tau_D$. For the ${}^4I_{11/2}(Er^{3+}) \rightarrow {}^6H_{5/2}(Dy^{3+})$ process, η_{ET} increases from 62.4% to 92.3%, and for the ${}^4I_{13/2}$

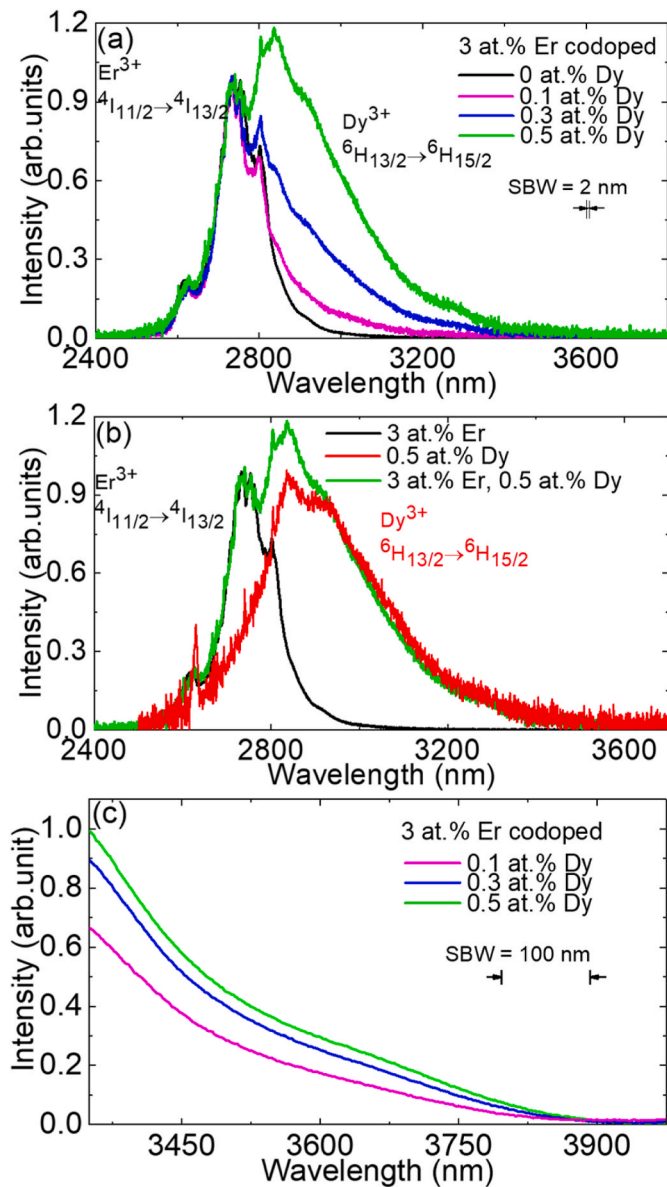


Fig. 6. Spectra of mid-infrared luminescence around 3 μm of (a) 3 at.% Er^{3+} , x at.% $\text{Dy}^{3+}:\text{CaF}_2$ crystals with $x = 0, 0.1, 0.3,$ and 0.5 ; (b) deconvolution of the emission spectrum of the 3 at.% Er^{3+} , 0.5 at.% Dy^{3+} codoped crystal using the spectra of the corresponding singly-doped CaF_2 crystals. $\lambda_{\text{exc}} = 960$ nm (Er), 1320 nm (Dy); (c) Long-wave multiphonon emission sideband of the ${}^6\text{H}_{13/2} \rightarrow {}^6\text{H}_{15/2}$ transition of Dy^{3+} ions beyond 3.3 μm .

$2(\text{Er}^{3+}) \rightarrow {}^6\text{H}_{11/2}(\text{Dy}^{3+})$ process – from 77.4% to 94.0%, when the Dy content changes from $x = 0.1$ at.% to 0.5 at.%.

4.7. 4.4- μm emission of Dy^{3+} ions and non-radiative relaxation

In the present work, we report on the first observation of 4.4- μm emission from Dy^{3+} ions in CaF_2 at room temperature. The SE cross-sections for the ${}^6\text{H}_{11/2} \rightarrow {}^6\text{H}_{13/2}$ transition were calculated using the F-L formula based on the measured luminescence spectrum, see Fig. 8. The peak σ_{SE} value reaches $1.36 \times 10^{-20} \text{ cm}^2$ at 4390 nm and the corresponding emission bandwidth (FWHM) is about 100 nm. The luminescence lifetime of this weak emission could not be directly measured at room temperature.

The luminescence lifetime of the ${}^6\text{H}_{11/2}$ manifold of 15 μs was estimated from the radiative lifetime calculated using the Judd-Ofelt theory, and the rate of multiphonon non-radiative relaxation extracted

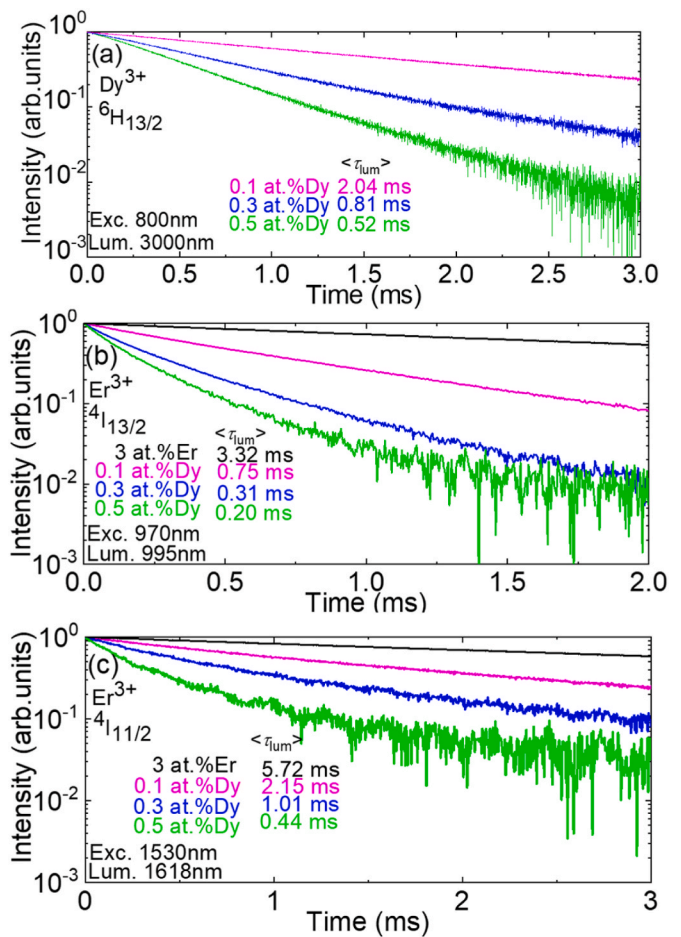


Fig. 7. (a-c) Luminescence decay curves from (a) ${}^6\text{H}_{13/2} \text{Dy}^{3+}$, (b,c) ${}^4\text{I}_{13/2}$ and ${}^4\text{I}_{11/2} \text{Er}^{3+}$ manifolds in 3 at.% Er, x at.% $\text{Dy}:\text{CaF}_2$ crystals with $x = 0.1, 0.3,$ and 0.5 , $\langle \tau_{\text{lum}} \rangle$ – mean luminescence lifetime. In (b,c), the data for 3 at.% Er singly-doped CaF_2 are added for comparison.

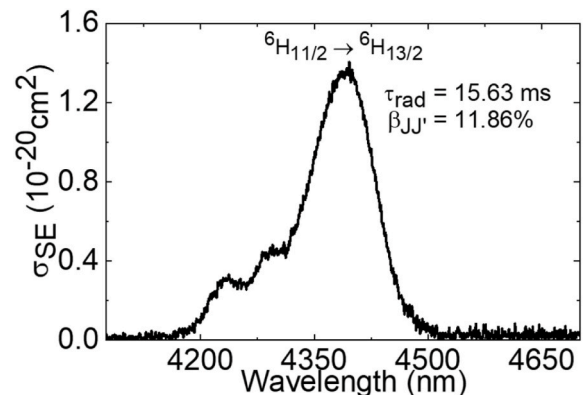


Fig. 8. Stimulated-emission (SE), σ_{SE} , cross-sections for the ${}^6\text{H}_{11/2} \rightarrow {}^6\text{H}_{13/2}$ transition of Dy^{3+} ions in CaF_2 .

form the exponential bandgap law, $(1/\tau_{\text{lum}}) = (1/\tau_{\text{rad}}) + W_{\text{NR}}$. The rate of multiphonon non-radiative relaxation is given by Ref. [47]:

$$W_{\text{NR}} = C e^{-\ln(\epsilon) \frac{\Delta E}{\hbar \nu_{\text{ph}}}}, \quad (3)$$

where C is a constant corresponding to the rate of NR relaxation in the limit of zero energy-gap ($\Delta E \rightarrow 0$), and ϵ is the ratio between the probabilities of m -phonon and $m-1$ -phonon relaxation. By fitting the

available experimental data on W_{NR} for various rare-earth ions (Ho^{3+} [24], Er^{3+} [18] and Dy^{3+} – this work) with Eq. (3), we have obtained the following material parameters: $C = 1.3 \pm 0.6 \times 10^9 \text{ s}^{-1}$ and $\varepsilon = 6.3 \pm 0.7$, see Fig. 9(a). Due to the small energy gap between the ${}^6\text{H}_{11/2}$ and ${}^6\text{H}_{13/2}$ states of around 2500 cm^{-1} , the luminescence lifetime of ${}^6\text{H}_{11/2}$ Dy^{3+} manifold at room temperature is strongly affected by the multiphonon-assisted thermal quenching demonstrated by a sharp drop of the lifetime from ms-range for τ_{rad} to μs -range for τ_{lum} . The non-radiative path from the ${}^6\text{H}_{11/2}$ manifold can be suppressed by cooling the $\text{Dy}^{3+}:\text{CaF}_2$ crystal to cryogenic temperatures.

Following the approach of Johnson and Guggenheim [14], we compared the rates of spontaneous dipole radiative transitions of rare-earth ions in CaF_2 – which scale with frequency as $\sim \nu^3$ – with the non-radiative relaxation rates, which decrease approximately exponentially, $\sim e^{-\nu}$, as shown in Fig. 9(b). At $\sim 3 \mu\text{m}$, the room-temperature non-radiative decay is less than two orders of magnitude higher than the radiative rate, still permitting laser operation despite a reduced luminescence quantum efficiency. At $4.4 \mu\text{m}$, this difference increases to nearly four orders of magnitude, necessitating cryogenic cooling to suppress phonon-assisted relaxation.

A potential laser scheme based on the ${}^6\text{H}_{11/2} \rightarrow {}^6\text{H}_{13/2}$ transition of Dy^{3+} ions would be self-terminating due to the significantly longer lifetime of the terminal level compared to that of the upper emitting level. This issue may be mitigated in two ways: i) by employing cascade lasing at both $3 \mu\text{m}$ and $4.4 \mu\text{m}$ to depopulate the intermediate ${}^6\text{H}_{13/2}$ manifold, or ii) by codoping with Tb^{3+} ions, which facilitate quenching of the ${}^6\text{H}_{13/2}$ lifetime. Notably, a similar approach has been successfully implemented in yellow Dy^{3+} lasers [48,49], where the non-negligible

lifetime of the terminal laser level is undesirable.

It is instructive to compare the $4.4 \mu\text{m}$ emission properties of Dy^{3+} ions in CaF_2 with those of the state-of-the-art mid-infrared-emitting Dy^{3+} -doped crystal KPb_2Cl_5 . For this compound, a stimulated-emission cross-section of $0.6 \times 10^{-20} \text{ cm}^2$ at $4.25 \mu\text{m}$ has been reported, although the polarization state was not specified. The ${}^6\text{H}_{11/2}$ luminescence lifetime τ_{lum} is 6.94 ms , corresponding to a luminescence quantum efficiency of 78% [50]. KPb_2Cl_5 is hygroscopic, and the growth of large-volume crystals with optical quality remains challenging; moreover, the material contains lead.

4.8. Discussion: Dy^{3+} ion in fluorite-type crystals

Within the family of cubic fluorite-type crystals MF_2 , the strontium and barium ($\text{M} = \text{Sr}$ and Ba) counterparts of calcium fluorides are known for yet lower phonon energies. We have directly compared the $3 \mu\text{m}$ emission properties (spectra and lifetimes) of calcium, strontium, and barium fluorides doped with $0.5 \text{ at.}\%$ Dy^{3+} . This concentration represents the lower limit at which the activator ions form clusters, thereby avoiding excessive luminescence quenching while enabling a meaningful comparison of multiphonon non-radiative relaxation processes as a function of host phonon energy. At lower doping levels, the contribution from isolated ions becomes significant.

Fig. 10(a) shows the Raman spectra of Dy^{3+} -doped CaF_2 , SrF_2 and BaF_2 crystals. At the center of the Brillouin zone $\Gamma(\mathbf{k} = 0)$, the factor group theory [51] predicts three optical phonon modes, i.e., a doubly degenerate IR-active mode $\text{TO } T_{1u}$, a triply degenerate Raman-active mode T_{2g} , and an IR-active nondegenerate mode $\text{LO } T_{1u}$. The frequency of the T_{2g} progressively decreases in the $\text{M} = \text{Ca} \rightarrow \text{Sr} \rightarrow \text{Ba}$ series, measuring $321.5 \rightarrow 284.0 \rightarrow 244.9 \text{ cm}^{-1}$, suggesting a decreased non-radiative path.

The spectra of $3\text{-}\mu\text{m}$ luminescence from these crystals, measured under identical excitation and detection conditions, are compared in Fig. 10(b). In the $\text{M} = \text{Ca} \rightarrow \text{Sr} \rightarrow \text{Ba}$ series, the spectra progressively narrower and become more structured due to the weaker crystal field and decreased diversity of the cluster geometries, while at the same time the luminescence intensity increases. This is in line with a progressive increase of the ${}^6\text{H}_{13/2}$ lifetime, $\tau_{lum} = 0.63 \text{ ms}$ ($\text{Dy}^{3+}:\text{SrF}_2$) and 1.13 ms ($\text{Dy}^{3+}:\text{BaF}_2$), Fig. 10(c). The corresponding rates of non-radiative relaxation W_{NR} are $1.57 \times 10^3 \text{ s}^{-1}$ and $0.86 \times 10^3 \text{ s}^{-1}$ and they progressively decrease in the $\text{M} = \text{Ca} \rightarrow \text{Sr} \rightarrow \text{Ba}$ series.

5. Conclusions

To conclude, calcium fluoride crystals doped with Dy^{3+} ions are promising gain media for broadly tunable lasers operating above $3 \mu\text{m}$ (on the ${}^6\text{H}_{13/2} \rightarrow {}^6\text{H}_{15/2}$ transition), i.e., outside the range of structured water vapor absorption being a severe limiting factor for mid-infrared Erbium lasers, e.g., those employing $\text{Er}^{3+}:\text{CaF}_2$ crystals. The $\text{Er}^{3+}, \text{Dy}^{3+}$ codoping scheme resolves the limitation of unconventional pump wavelengths for Dy^{3+} : efficient excitation of $3 \mu\text{m}$ luminescence is demonstrated by InGaAs diode laser pumping. Codoped crystals exhibit high energy-transfer efficiencies (exceeding 90% , estimated from the shortening of the donor (Er^{3+}) luminescence lifetimes) promoted by the proximity of donor and acceptor (Dy^{3+}) ions forming rare-earth clusters. The $\text{Er}^{3+}, \text{Dy}^{3+}$ codoped CaF_2 crystals also feature increased emission bandwidth due to the spectrally combined luminescence profiles of both active ions. This represents an additional advantage as compared to the alternative $\text{Tm}^{3+}, \text{Dy}^{3+}$ codoping scheme exploited earlier [52].

In the challenging spectral range of 3 to $4 \mu\text{m}$, two emission mechanisms are revealed: i) a phonon-assisted (Stokes) emission related to the ${}^6\text{H}_{13/2} \rightarrow {}^6\text{H}_{15/2}$ transition sharing the same upper laser level (${}^6\text{H}_{15/2}$) lifetime, and ii) a relatively weak emission arising from the ${}^6\text{H}_{11/2} \rightarrow {}^6\text{H}_{13/2}$ electronic transition which is in a strong competition with a non-radiative relaxation path. The potential of fluorite-type crystals for emission at $4.4 \mu\text{m}$ could be further revealed when employing SrF_2 and

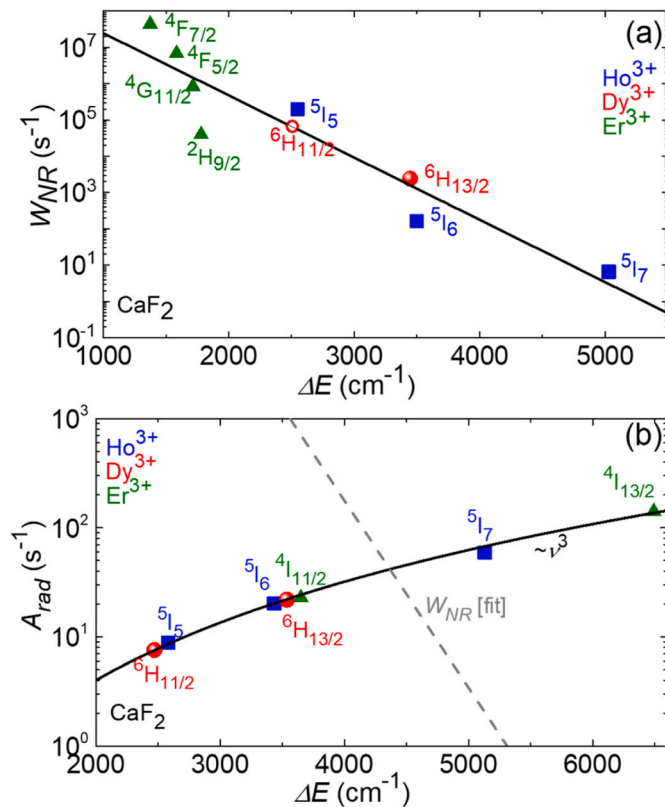


Fig. 9. Probabilities of radiative and non-radiative transitions of rare-earth ions in CaF_2 : (a) non-radiative relaxation rate W_{NR} vs. the energy gap to the next lower-lying level ΔE : symbols – experimental data derived from the luminescence and radiative lifetimes, line – their fit using Eq. (3); (b) probabilities of spontaneous radiative transitions A_{rad} from the emitting manifolds: symbols – values obtained using the Judd-Ofelt theory, curve – their fit using a ν^3 dependence expected for dipolar transitions.

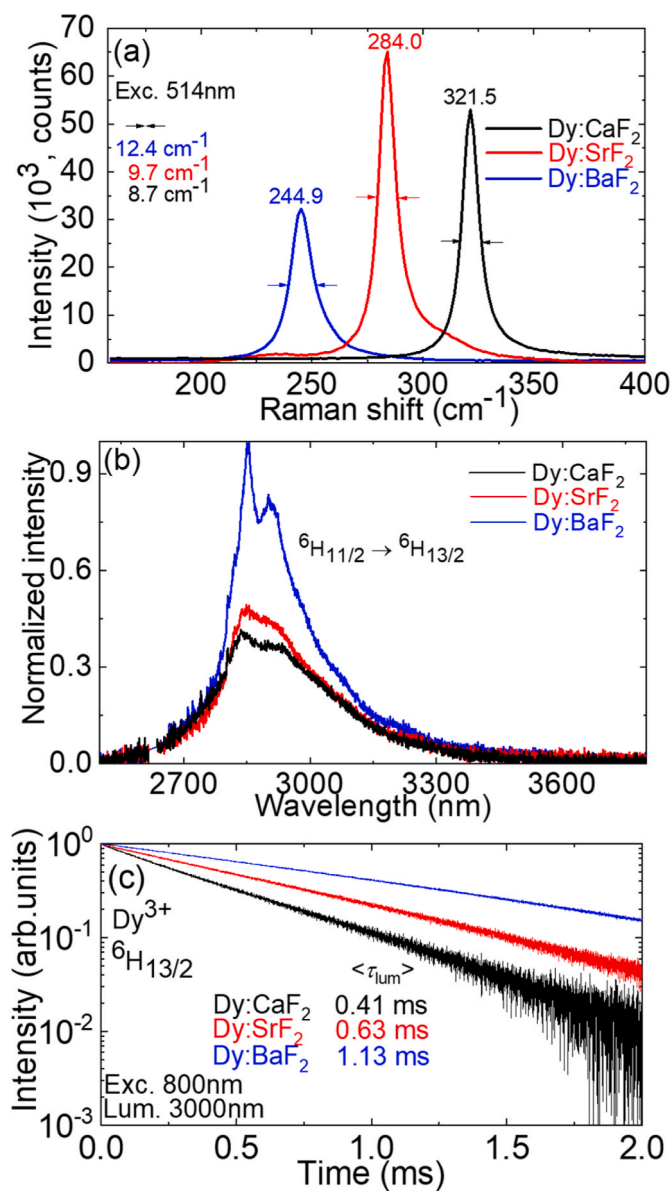


Fig. 10. Comparison of vibronic and mid-infrared emission properties of fluorite-type crystals 0.5 at.% Dy:MF₂ (M = Ca, Sr and Ba): (a) Raman spectra, $\lambda_{exc} = 514$ nm, numbers – peak frequencies and Raman peak widths (FWHM); (b) luminescence spectra at 3 μm (in scale); (c) Luminescence decay curves from the ${}^6\text{H}_{13/2}$ Dy³⁺ manifold, $\langle\tau_{lum}\rangle$ – mean luminescence lifetime.

BaF₂ host matrices for Dy³⁺ doping.

Given the high solubility of rare-earth activators in fluorite-type crystals, the study of heavily Dy³⁺-doped MF₂ systems is of particular interest for identifying the mechanisms responsible for luminescence quenching for the ${}^6\text{H}_{13/2}$ manifold, possibly involving energy migration. The corresponding lifetime may also be influenced by the increasing structural complexity of rare-earth ion clusters at high doping levels.

CRedit authorship contribution statement

Ngoc Quynh Hoa Nguyen: Investigation, Writing – original draft. **Elena Dunina:** Formal analysis, Methodology. **Liudmila Fomicheva:** Formal analysis, Methodology. **Alexey Kornienko:** Data curation, Methodology, Software. **Pavel Popov:** Investigation, Methodology. **Sergei Kuznetsov:** Conceptualization, Investigation, Resources. **Abdelmjid Benayad:** Investigation. **Patrice Camy:** Funding acquisition, Supervision. **Alain Braud:** Funding acquisition, Supervision,

Writing – review & editing. **Pavel A. Loiko:** Writing – review & editing, Methodology, Supervision.

Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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Data availability

Data will be made available on request.

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